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Sulphuric acid for industrial use — Determination of chlorides content — Potentiometric method**AMENDMENT 1**

Amendment 1 to International Standard ISO 2877 was drawn up by Technical Committee ISO/TC 47, *Chemistry*. It was submitted directly to the ISO Council, in accordance with clause 6.12.1 of the Directives for the technical work of ISO.

Page 3

Annex : The results given in the example were obtained using a platinum electrode. As it was subsequently agreed to use a silver electrode (see 5.1.4), the example should be related to the results obtained using a silver electrode, even if the analytical result is essentially the same. Hence, replace the table by the following :

Volume of silver nitrate solution <i>V</i>	Potential <i>E</i>	$\Delta_1 E$	$\Delta_2 E$
ml	mV		
4,8	125	9	+ 16
4,9	134	25	+ 91
5,0	159	116	– 67
5,1	275	49	
5,2	324		
$V_{EQ} = 5,1 + 0,1 \times \frac{91}{91 + 67} = 5,157$			

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